

Things to look for BEFORE answering an equilibrium Q		
Stressor	Question	What does it tell us?
Increase or decrease [] products or reactants	Which phase?	<ul style="list-style-type: none"> Gas, aqueous - change things Solid, Liquid – DON'T CHANGE ANYTHING!
Increase or decrease T	Endo or exo?	<ul style="list-style-type: none"> Endo = absorbed, so it is a REACTANT Exo = released, so it is a PRODUCT
Increase or decrease total Pressure <i>(Same as Δ in V or # of moles of gas)</i>	How many moles of GAS are on each side of the equation?	<ul style="list-style-type: none"> Increase pressure = move to side with FEWER moles of gas Decrease pressure = move to side with MORE moles of gas

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